

Supramolecular structures of 2-cyano-3-dimethylamino- *N*-(4-methylphenyl)acrylamide and 2-cyano-3-dimethylamino- *N*-(2-methoxyphenyl)acrylamide

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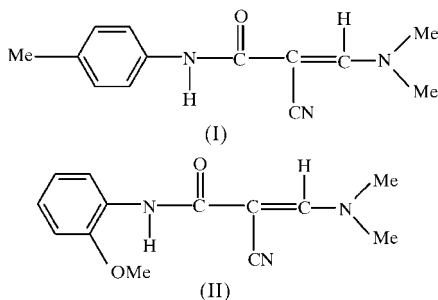
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In the title compounds, C₁₃H₁₅N₃O, (I), and C₁₃H₁₅N₃O₂, (II), the dihedral angles between the planes of the phenyl ring and the amide group are 4.1 (1) and 20.7 (1)°, respectively. The molecules adopt a fully extended conformation, aided by intramolecular interactions. The molecular structures of (I) and (II) display different crystal packing and hydrogen-bonding networks.

Comment

As part of our study of conformational analysis, crystallographic work on *N*-aromatic amide derivatives has been undertaken. These derivatives are analogs of the active metabolites of the immunosuppressive drug leflunomide, which are known to act, in part, by inhibiting the tyrosine



kinase epidermal growth-factor receptor (EGFR; Mattar *et al.*, 1993). EGFR is a membrane-associated tyrosine kinase, which serves as an endogenous negative regulator of apoptosis in

breast cancer cells (Uckun *et al.*, 1998). The present study reports the structures of two *N*-acrylamide compounds, (I) and (II) (Figs. 1 and 2), in order to examine the effects of substituents on the hydrogen-bonding systems and on the crystal structures.

The dihedral angle between the planes of the phenyl rings and the amide groups are 4.1 (1) and 20.7 (1)° for (I) and (II), respectively. In both compounds, the geometry of the amide group is comparable to that of similar groups in acetanilides (Haisa *et al.*, 1977). The C10–C11 and C11–N12 bond lengths (Tables 1 and 3) agree with expected Csp^2-Csp [1.431 (14) Å] and $Csp-N$ [1.136 (10) Å] bond lengths, respectively (Allen *et al.*, 1987). Similar observations have been noted for the crystal structures of the leflunomide metabolite analogs (Ghosh *et al.*, 1999; Ghosh & Uckun, 1999) and for an acrylamide derivative (Ompraba *et al.*, 2003). In (II), the C2–O17–C18 angle [118.3 (1)°] is close to that expected for sp^2 hybridization of atom O17. The distortion and enlargement of the C6–C1–N7, C1–N7–C8 and N7–C8–O9 angles from the trigonal value (120°) are due to intramolecular C6–H6···O9 hydrogen bonds (Tables 2 and 4). In both (I) and (II), the cyanoacrylamide side chain is planar, with π -conjugation along the chain causing variations of the bond distances with respect to localized double and single bonds. The C1–N7–C8–C10 torsion angle does not differ significantly between (I) and (II) [–178.9 (1) and 176.6 (1)°, respectively], whereas the C6–C1–N7–C8 angle differs substantially [–3.7 (3) and –20.2 (2)°, respectively], indicating that the large twist around the C1–N7 bond in (II) is due to an intramolecular N7–H7···O17 hydrogen bond. This hydrogen bond determines the orientation of the

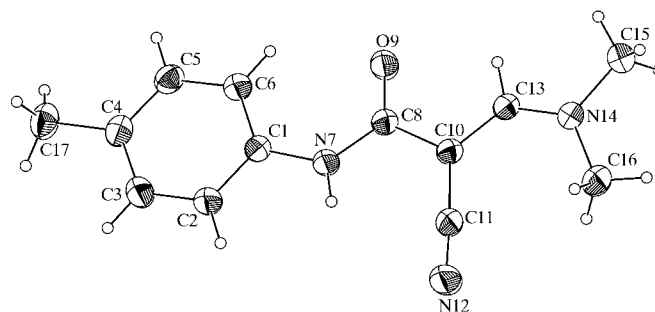


Figure 1

The molecular structure of (I), showing displacement ellipsoids at the 35% probability level.

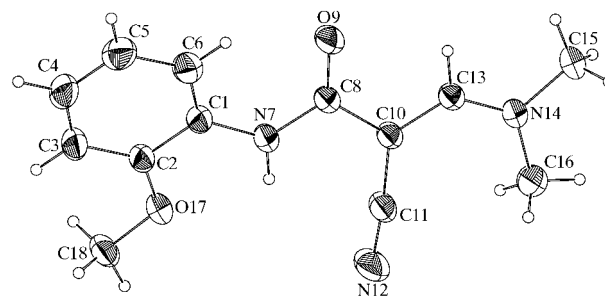


Figure 2

The molecular structure of (II), showing displacement ellipsoids at the 35% probability level.

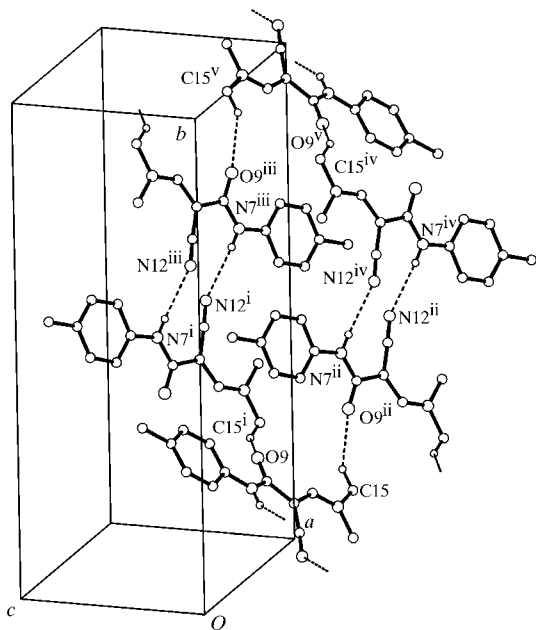


Figure 3

A view of the discrete hexamer formed by the molecules of (I). [Symmetry codes: (i) $x, \frac{1}{2} - y, \frac{1}{2} + z$; (ii) $x, \frac{1}{2} - y, -\frac{1}{2} + z$; (iii) $2 - x, \frac{1}{2} + y, \frac{1}{2} - z$; (iv) $2 - x, \frac{1}{2} + y, -\frac{1}{2} - z$; (v) $2 - x, 1 - y, -z$.]

methoxy group [C1—C2—O17—C18 = 179.5 (1)°], which is coplanar with the phenyl ring.

The supramolecular structures of (I) and (II) are completely different. In (I), the symmetry-related molecules are linked together head-to-tail *via* N7—H7···N12($2 - x, -y, z$) hydrogen bonds to form a dimer comprising an $R_2^2(12)$ ring (Bernstein *et al.*, 1995). The dimers at ($x, \frac{1}{2} - y, \frac{1}{2} + z$) and ($2 - x, \frac{1}{2} + y, \frac{1}{2} - z$) [center of symmetry at ($1, \frac{1}{2}, \frac{1}{2}$)], and ($x, \frac{1}{2} - y, -\frac{1}{2} + z$) and ($2 - x, \frac{1}{2} + y, -\frac{1}{2} - z$) [center of symmetry at ($1, \frac{1}{2}, -\frac{1}{2}$)], are further linked by symmetry-related C—H···O hydrogen bonds, linking atom O9 at (x, y, z) with atom H15A at ($x, \frac{1}{2} - y, \frac{1}{2} + z$), atom H15A at (x, y, z) with atom O9 at ($x, \frac{1}{2} - y, -\frac{1}{2} + z$), atom O9 at ($2 - x, 1 - y, -z$) with atom H15A at ($2 - x, \frac{1}{2} + y, -\frac{1}{2} - z$), and atom H15A at ($2 - x, 1 - y, -z$) with atom O9 at ($2 - x, \frac{1}{2} + y, \frac{1}{2} - z$), respectively, thus forming an $R_6^6(36)$ ring. Hence, a discrete hexamer is formed with the center of symmetry at ($1, \frac{1}{2}, 0$) (Fig. 3). The result is a two-dimensional layer, which runs along the *bc* plane (Fig. 4). In (II), an intramolecular N7—H7···O17 hydrogen bond forms a five-membered ring. A *C*(8) motif is formed *via* a C4—H4···O9($1 - x, \frac{1}{2} + y, \frac{3}{2} - z$) hydrogen

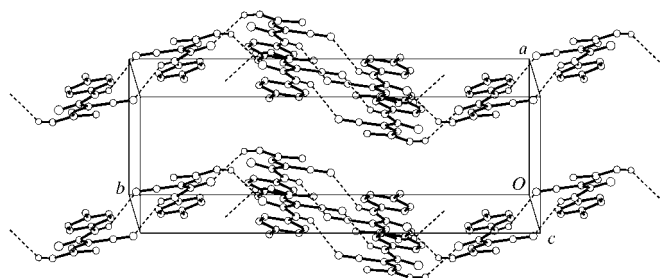


Figure 4

The molecular packing of (I), viewed along the *c* axis.

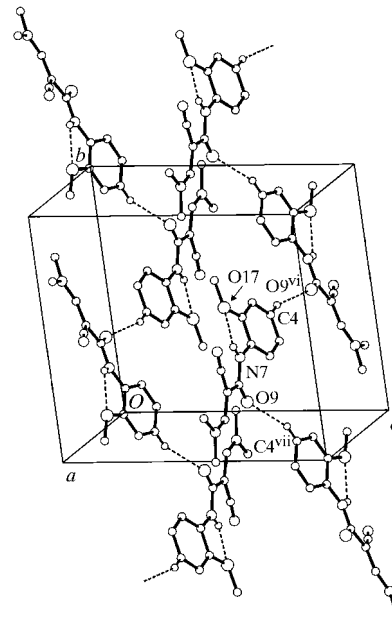


Figure 5

A view of the crystal structure of (II), viewed along the *a* axis, showing two antiparallel *C*(8) chains. [Symmetry codes: (vi) $1 - x, \frac{1}{2} + y, \frac{3}{2} - z$; (vii) $1 - x, -\frac{1}{2} + y, \frac{3}{2} - z$.]

bond, creating a chain that runs parallel to the *ab* plane. Two such antiparallel chains are shown in Fig. 5. A C—H··· π interaction is also observed in (II) (Table 4). The H7···C11 distances, *viz.* 2.34 Å in (I) and 2.28 Å in (II), are short as a result of the positive charge on atom H7 and the negative charge on atom C11.

Experimental

Substituted *N*-arylcyanacetamide (0.005 mol) was dissolved in dimethylformamide (6 ml) and kept under ice-cold conditions. To this solution, POCl₃ (1.4 ml, 0.015 mol) was added slowly with stirring. The reaction mixture was allowed to reach room temperature and was stirred for 3–4 h. The residue was then poured on to crushed ice and neutralized with NaOH (10%), and the crude product was filtered, washed with water and dried. Finally, the compound was purified by recrystallization using an ethyl acetate–petroleum ether mixture [m.p. 445 and 425 K for (I) and (II), respectively].

Compound (I)

Crystal data

C₁₃H₁₅N₃O
M_r = 229.28
 Monoclinic, $P2_1/c$
a = 7.5846 (3) Å
b = 22.4477 (10) Å
c = 7.5989 (3) Å
 β = 106.306 (1)°
V = 1241.72 (9) Å³
Z = 4

D_x = 1.226 Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 4951 reflections
 θ = 1.8–28.3°
 μ = 0.08 mm⁻¹
T = 293 (2) K
 Slab, pale yellow
 0.48 × 0.40 × 0.16 mm

Data collection

Siemens SMART CCD area-detector diffractometer
 ω scans
 8417 measured reflections
 3022 independent reflections
 2242 reflections with $I > 2\sigma(I)$

R_{int} = 0.037
 θ_{max} = 28.3°
 h = -10 → 8
 k = -29 → 27
 l = -10 → 8

Table 1
Selected geometric parameters (Å, °) for (I).

C1—N7	1.415 (2)	C11—N12	1.148 (2)
N7—C8	1.364 (2)	C13—N14	1.320 (2)
C8—O9	1.228 (2)		
C6—C1—N7	124.6 (2)	N14—C13—C10	132.0 (2)
C2—C1—N7	117.0 (1)	C13—N14—C16	124.1 (2)
C8—N7—C1	128.4 (1)	C13—N14—C15	120.4 (2)
O9—C8—N7	122.5 (2)	C16—N14—C15	115.5 (2)
C6—C1—N7—C8	−3.7 (3)	N7—C8—C10—C13	−170.2 (1)
C1—N7—C8—C10	−178.9 (1)	C11—C10—C13—N14	0.1 (3)

Table 2
Hydrogen-bonding geometry (Å, °) for (I).

D—H...A	D—H	H...A	D...A	D—H...A
C6—H6...O9	0.93	2.27	2.864 (2)	121
C13—H13...O9	0.93	2.34	2.748 (2)	106
N7—H7...N12 ^{viii}	0.86	2.50	3.228 (2)	143
C15—H15A...O9 ⁱⁱ	0.96	2.57	3.360 (2)	139

Symmetry codes: (ii) $x, \frac{1}{2} - y, -\frac{1}{2} + z$; (viii) $2 - x, -y, -z$.

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.057$
 $wR(F^2) = 0.169$
 $S = 1.04$
 3022 reflections
 157 parameters
 H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0849P)^2 + 0.2309P]$$

where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.19 \text{ e } \text{Å}^{-3}$
 $\Delta\rho_{\min} = -0.15 \text{ e } \text{Å}^{-3}$

Compound (II)

Crystal data

$\text{C}_{13}\text{H}_{15}\text{N}_3\text{O}_2$
 $M_r = 245.28$
 Monoclinic, $P2_1/c$
 $a = 7.5141 (3) \text{ Å}$
 $b = 12.7580 (6) \text{ Å}$
 $c = 13.9381 (6) \text{ Å}$
 $\beta = 92.795 (1)^\circ$
 $V = 1334.58 (10) \text{ Å}^3$
 $Z = 4$

$D_x = 1.221 \text{ Mg m}^{-3}$
 Mo $K\alpha$ radiation
 Cell parameters from 5915 reflections
 $\theta = 2.2\text{--}28.3^\circ$
 $\mu = 0.09 \text{ mm}^{-1}$
 $T = 293 (2) \text{ K}$
 Block, pale yellow
 $0.48 \times 0.46 \times 0.42 \text{ mm}$

Data collection

Siemens SMART CCD area-detector diffractometer
 ω scans
 9025 measured reflections
 3278 independent reflections
 2425 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.041$
 $\theta_{\max} = 28.3^\circ$
 $h = -9 \rightarrow 9$
 $k = -16 \rightarrow 16$
 $l = -12 \rightarrow 18$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.048$
 $wR(F^2) = 0.152$
 $S = 1.05$
 3278 reflections
 166 parameters
 H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0825P)^2 + 0.0796P]$$

where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.19 \text{ e } \text{Å}^{-3}$
 $\Delta\rho_{\min} = -0.14 \text{ e } \text{Å}^{-3}$

Table 3
Selected geometric parameters (Å, °) for (II).

C1—N7	1.407 (2)	C11—N12	1.147 (2)
N7—C8	1.368 (2)	C13—N14	1.316 (2)
C8—O9	1.228 (2)		
C6—C1—N7	124.6 (1)	N14—C13—C10	130.8 (1)
N7—C1—C2	115.7 (1)	C13—N14—C16	123.8 (1)
C8—N7—C1	128.4 (1)	C13—N14—C15	120.1 (1)
O9—C8—N7	122.2 (1)	C16—N14—C15	116.1 (1)
C6—C1—N7—C8	−20.2 (2)	C11—C10—C13—N14	3.4 (2)
C1—N7—C8—C10	176.6 (1)	C1—C2—O17—C18	179.5 (1)
N7—C8—C10—C13	177.3 (1)		

Table 4
Hydrogen-bonding geometry (Å, °) for (II).

Cg is the centroid of the C1—C6 ring.

D—H...A	D—H	H...A	D...A	D—H...A
C6—H6...O9	0.93	2.38	2.910 (2)	116
C13—H13...O9	0.93	2.36	2.756 (2)	106
N7—H7...O17	0.86	2.22	2.596 (1)	106
C4—H4...O9 ^{vi}	0.93	2.57	3.452 (2)	158
C18—H18C...Cg ^{ix}	0.96	2.75	3.551 (1)	141

Symmetry codes: (vi) $1 - x, \frac{1}{2} + y, \frac{3}{2} - z$; (ix) $1 - x, 1 - y, 1 - z$.

For both compounds, data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; structure solution: *SHELXS97* (Sheldrick, 1997); structure refinement: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ZORTEP* (Zsolnai, 1997) and *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97* and *PARST* (Nardelli, 1995).

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: NA1594). Services for accessing these data are described at the back of the journal.

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